# **OPTIMO ATCZ291**

Optimization of sustainable sulfuric acid production for (bio)leaching processes in the waste sector



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**Scientific Partners** 



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### **OPTIMO ATCZ291**

Optimization of sustainable sulfuric acid production for (bio)leaching processes in the waste sector



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#### **Sulfuric acid**

- Sulfuric acid is the world's largest bulk chemical, but production high energy requirements
- On the other hand, elemental sulfur occurs in large amounts as a by-product

#### Aim

- Demonstrate the high potential of extremely acidophilic, sulfur-oxidizing bacteria
- An efficient biogenic sulfuric acid production
- Bioelectrochemical metal recovery





#### opäischer Fonds für regionale Entwicklung

#### **Circular economy**

 Use of the produced sulfuric acid for (bio)leaching of secondary raw materials

#### Partner OPTIMO - Interreg ATCZ291





**K1-MET, Linz:** Lead partner and experiments with thermophilic bacteria and bioelectrochemical zinc recovery



**BOKU, Institute of Environmental Biotechnology, Tulln:** experiments with mesophilic bacteria and heap bioleaching



**MU, Masaryk University, Brno:** Metal analytics, characterization of the microbial communities, kinetic analysis

Project budget:	191 949.09 € (85% funding rate)
Project duration:	01.01.2022 – 31.12.2022 (one year)
Funded by:	European Regional Development Fund





### Selection of suitable bacteria Bioleaching



BOKU: mesophilic bacteria: Acidithiobacillus thiooxidans, A. caldus and

environmental sample (lake in Czech Republic)

• K1-MET: thermophilic bacteria: Sulfobacillus acidophilus, S. thermosulfidooxidans

Incubation with elemental sulfur

Screening according:

- Produced acidity
- Sulfur conversion
- Optical density







Incubation with elemental sulfur

### Selection of potential waste streams Biogenic Sulfuric Acid as a leaching agent



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Transparent pharmaceutical blisters









White pharmaceutical blisters





#### Printed circuit boards (PCBs)



### Selection of suitable parameters Bioleaching at BOKU



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- 1.5 10% (w/v) elemental sulfur
- 250 rpm
- 30°C
- 100 L/h aeration
- Minimal Medium (DSM35)
- Influence of CO<sub>2</sub> and O<sub>2</sub>



Laboratory Setup at BOKU (©IFA Tulln)

More about biogenic sulfur production at BOKU today...
 Biogenic sulfuric acid production and application for metal bioleaching from different waste streams
 Dr. Klemens Kremser





## **Bioelectrochemical Experiments** Zinc recovery at K1-MET

- □ Set up of a microbial electrolysis cell (bioanode and cathode)
- Testing Zn recovery from synthetic solution
  - □ >90% Zn recovery
  - □ Impact of cathode material (titanium, graphite)
  - □ Impact of flushing the cathode solution at Zn recovery



 $\Box$  Sampling of electroactive biofilm  $\rightarrow$  Analysis of microbial community (MU)

Microbial electrolysis cell









#### Metal recovery from bioleachate - Method



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Setup	
Cathode	<b>220 mL diluted bioleachate</b> Graphite Aerobic (without catholyte flushing)
Anode	220 mL Cheng medium (pH 7) Pretreated carbon felt electrode Anaerobic
Membrane	Proton exchange membrane (PEM)
Microbes	Mixed culture (inoculum sewage sludge)
Substrate	2x per week, synthetic wastewater (glucose, acetate, peptone, yeast extract)
Temperature	Room temperature (RT)
Monitoring	pH, current flow, anode/cathode potential, COD removal rate, Zn recovery rate
Applied potential	Anode: -100mV vs. Ag/AgCI (3M NaCI)







## **Characterization of microbial communities**

(Meta)genomic analysis at Masaryk University

- Methods and Instruments
  - MiniSeq System (Illumina) next-generation sequencing technology
    >> monitoring all prokaryotes
  - Quantitative PCR (QPCR)

>> monitoring of the selected species or metabolism markers



https://www.illumina.com/



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- Study of the limitation of sulfur and iron oxidation by gases such as O<sub>2</sub> and CO<sub>2</sub>
- Model bioleaching microorganism Acidithiobacillus ferrooxidans
- Impact on mixing and aeration requirements in operational processes



• More today ....

The production of biogenic sulfuric acid is limited by the sulfur substrate and potentially by carbon dioxide *Dr. Martin Mandl* 

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